

Single-Molecule Detection in Dilute Liquids and Live Cells without Immobilization or Significant Hydrodynamic Flow at Room Temperature or under Physiological Conditions: Too Much Thermodynamic Jitter

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ABSTRACT

Although the experimental attempts to measure a single molecule/particle, i.e., an individual molecule/particle, in dilute liquids or live cells without immobilization on a surface or hydrodynamic flow at room temperature or under physiological conditions have failed so far, this failure spurred the theory on Brownian molecular motion based on the stochastic nature of diffusion. This new physical theory for the quantifying the thermodynamic jitter of molecules/particles is inspiring for many and forms the theoretical basis of single-molecule biophysics and biochemistry,

which underlies the stochastic nature of diffusion. Theoretical considerations for analyzing mobility data are summarized. Measuring the individual molecule or the individual particle is considered among the most challenging trends of research in spectroscopy, microscopy and nanoscopy.

Keywords: Diffusion, Thermodynamic jitter, Nanoscopy, Ergodicity

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INTRODUCTION

The transport and binding of molecules to specific targets are necessary for the assembly and function of ordered supramolecular structures in cells. Analysis of these processes *in vitro* and *in vivo* is central in life sciences including pharmaceutical sciences. In cellular compartments transport is frequently driven by diffusion, which tends to balance concentration gradients. Mobility has been studied during the last decades by different experimental technologies, mainly based on fluorescence techniques with confocal or wide-field optics including laser scanning microscopy. Thus, time-resolved protein mobility data are related to images of cellular structures from fluorescence confocal laser scanning microscopy to identify localization-specific dynamics and interactions of a fluorescently labeled species. The spatial resolution of the mobility and interaction analysis in super-resolution microscopy is below the diffraction limit and it is not anymore limited by the diffraction-limited size of the excitation volume (observation/detection volume) in a confocal fluorescence microscopy/spectroscopy setup. For many of these approaches, the Green Fluorescent Protein (GFP) and its spectral relatives are particularly well suited in the intracellular environment due to their high-fluorescence yield, their photobleaching properties (Harms GS, *et al.*, 2001), and because they can be fused to proteins *in vivo* (White J and Stelzer E, 1999). As a result of these technological and biotechnological developments, single-molecule studies in liquids and live cells have entered the scene a couple years ago. Such studies are related to measurements of individual molecules (individual particles), but they are not the same. Fortunately, in cases of diffusion-controlled reactions and systems, respectively, there are physical criteria (see below the criteria 1 to 4) to make the decision. The figures summarize the most important theoretical results for measuring just one molecule at a time (individual molecule/particle) in dilute liquids at room temperature or live

cells under physiological conditions without immobilization on a surface or hydrodynamic flow, i.e., the self-same molecule/particle. Here, we contribute in the field for the benefit of researchers, for example mainly in pharmacy, who have no deep experience with the single-molecule detection in liquids and live cells without immobilization on a surface or without significant hydrodynamic flow.

LITERATURE REVIEW

The main idea of single-molecule detection without immobilization on a surface or without hydrodynamic flow in liquids at room temperature or live cells under physiological conditions is simple. A single-molecule/single-particle experiment is an experiment that investigates the properties of individual molecules/particles. Two means are of great importance. First, only one probe molecule/particle should diffuse in the excited volume, and second, the light emitted from this single molecule/particle, that is the individual molecule/particle or the selfsame molecule/particle, should be distinguishable from the experimental noise. So far so simple, but it becomes much more difficult when such a scenario is quantified in terms of the temporal motion (time resolution) of individual molecules/individual particles. The thermodynamic concept of diffusion (fluctuations, thermodynamic jitter) due to the same molecule, that is the individual molecule, was developed by Földes-Papp Z, 2007; Földes-Papp Z, 2007; Földes-Papp Z, 2006; Földes-Papp Z, *et al.*, 2005; Földes-Papp Z, 2013) and from the mathematical core to the physical theory of single-molecule biophysics and biochemistry based on the stochastic nature of diffusion (Földes-Papp Z, 2015; Földes-Papp Z, *et al.*, 2021) and it was considered pioneering work (Digman MA and Gratton E, 2011). The conceptual and theoretical basis of thermodynamic jitter has been established in series of publications.

Physical criteria are required to decide whether it is a single mol-

ecule/a single particle (an individual molecule/an individual particle) or not, as we illustrate in *Figure 1*. These criteria (criteria 1 to 3) result in the time-resolution of an individual molecule/individual particle in optical microscopy/nanoscopia (that is the criterion 4), which covers the most important families of dynamical systems with randomness, that means here molecule number fluctuations are currently of interest. Main theoretical results for measuring just one molecule/particle at a time (individual molecule/particle) in liquids or live cells without immobilization or hydrodynamic flow, i.e. the self-same molecule/particle (Földes-Papp Z, 2013; Földes-Papp Z, 2021; Baumann G and Földes-Papp Z, 2022).

DISCUSSION

Stochastic thermodynamics rules the physical formulation of the single-molecule time resolution (T_m) which can be explained by the below mentioned equation-

$$T_m = \tau_{\text{dif}}(t) / c_m \cdot \Delta V \cdot K$$

Where, K stands for the proportionality factor of the equation with $K = e \cdot (c_m \cdot N_A \cdot \Delta V) / N_A$

Here e = Natural exponential function

T_m = Single molecule time resolution, i.e., the meaningful time of measuring just the same molecule in dilute solutions and live cells without immobilization or hydrodynamic flow.

$\tau_{\text{dif}}(t)$ = Diffusion time of the molecule

c_m = Molar concentration of molecules of the same kind in the bulk (bulk phase).

ΔV = Observation/detection volume

N_A = Avagadro constant number and it is defined as $N_A = 6.02214076 \times 10^{23} \text{ mol}^{-1}$

Obviously the dimension of T_m is the dimension of τ_{dif}

The number of molecules averaged over the measurement time T is gives as mentioned below-

$$N_{\text{Imax}} = T / T_m$$

The basic results are due to Markov processes which are the door to dynamical systems that fluctuate, for example in their number of molecules under observation. The probability of separating two individual molecules or two individual particles as independent molecular units during the measurement time is the criterion 3 as demonstrated by Földes-Papp Z, 2021 in the *Table 1* there under different experimental measurement conditions.

Looking deeper into this physical theory of single-molecule detection of one and the same molecule/particle (the individual molecule/particle) that is the theory of single molecule biophysics and biochemistry based on the stochastic nature of diffusion in dilute liquids and single live cells without immobilization on artificial or biological surfaces/membranes or without significant hydrodynamic flow is merely the way to understand what occurs during diffusion of an individual molecule/particle in the observation/detection volume embedded into a bulk phase such as liquids as well as live cell compartment or membrane (Földes-Papp Z and Baumann G, 2011; Baumann G, *et al.*, 2010). Let us now focus on this in a more broadly comprehensible way.

So, we have diffused to about 1 nanometer from another protein. Our encounter has begun, and in roughly a nanosecond we will either touch the surface of the other protein, or be repelled by it. What happens in this nanosecond occupied several publications. We got into the ergodic hypothesis where it was clear that 'single molecule' behavior should not be extrapolated to all molecule behavior (Földes-Papp Z and Baumann G, 2011). By simulation experiments we found the minimal variation if the number of

randomly selected single-molecule tracks is $N_{\text{Emax}} = 32$. All other values of N_{E} deliver only a local minimum instead of a global minimum. The graphs in *Figure 1* also showed that the variation approaches a stable value if N_{E} approaches large values; i.e. only a small subpopulation of single molecules delivers the minimal variation (Földes-Papp Z and Baumann G, 2011). In other words, the most striking feature of performing ensemble averaging in sparse subpopulations of single molecules, however, is the same mean value obtained in an ergodic system that is a many molecule system, if the number of randomly selected single-molecule tracks is $N_{\text{Emax}} = 32$. Hence, broken ergodicity and unbroken ergodicity are not anymore distinguishable. In addition, when averaging procedures are carried out without knowing whether the underlying molecular system behaves in ergodic or non-ergodic ways, each measurement can be related to an ergodic or a non-ergodic behavior unless one is able to show the single-molecule fingerprint of non-ergodicity (Földes-Papp Z and Baumann G, 2011).

The single-molecule/particle fingerprint of non-ergodicity is the thermodynamic fingerprint (signature) of an individual molecule/particle in liquids or live cells without immobilization or without significant hydrodynamic flow (Földes-Papp Z, 2023). The novel physical theory presented offers a new way to understand the molecular behavior when single biomacromolecules are trapped in interactions with their neighboring ligands and reaction partner(s), respectively, in a crowded environment at room temperature or at physiological conditions in live cells and their cellular compartments (Baumann G and Földes-Papp Z, 2022). Therefore, single-molecule studies may be contrasted with measurements on an ensemble or bulk collection of molecules, where the individual behavior of molecules cannot be distinguished and only average characteristics can be measured. It clearly follows from these results that the biggest breakthrough in microscopy/nanoscopia and spectroscopy would be a breakthrough in sensitivity for measuring an individual single molecule (one and the same molecule) over several milliseconds to seconds and even minutes without immobilization on artificial surfaces or biological membranes as well as without significant hydrodynamics.

The measurement time for measuring just one single molecule (individual molecule, selfsame molecule) or one single particle (individual particle, selfsame particle) is a meaningful time in contrast to the measurement times for the many-molecule measurements. From the inspection of the mathematically derived and well experimentally founded physical relationships on the basis of stochastic translational diffusion (Baumann G and Földes-Papp Z, 2022), we can finally distinguish three types of meaningful times-

- The meaningful time as single molecule/single particle time- resolution discussed here in detail (Földes-Papp Z, 2013),
- The meaningful time as limits in measurement time that should not be exceeded in order to follow the same single molecule/same single particle with high probability in one, two or three dimensions also discussed above (*Table 1*) (Földes-Papp Z, 2021)
- The meaningful time as quantitative measure for the meaningful reentries of the same single molecule/same single particle in the observation/detection volume (Földes-Papp Z, 2007).

The dimensions of the meaningful times are the dimensions of the diffusion times of the molecule/particle in liquids and live cells without immobilization or without significant hydrodynamic flow. The measurement of the individual molecule or the individual particle is considered one of the most demanding research trends in spectroscopy, microscopy and nanoscopia because individual molecules (oligonucleotides and single-stranded DNA sequences) were theoretically described in chemical oligonucleotide syntheses on solid supports after release from the solid phase as early as 1994 for the first time (Földes-Papp Z, *et al.*, 1994).

Table 1: Experimental conditions that must be set up in the “single-molecule experiments” to analyze the meaningful times as shown in the Figure 1 in liquids, living cells and artificial as well as biological membranes without immobilization and without hydrodynamic flow (Földes-Papp Z, 2007; Földes-Papp Z, et al., 2005; Földes-Papp Z, 2006; Földes-Papp Z, 2021)

Experimental criteria	Formulas
Criterion 1	$N \leq 1$
N is the absolute number of the specific molecules (labeled, studied) in the observation/detection volume	It is the Poisson probability of detecting single molecules of the same kind in the observation/detection volume
Criterion 2	$\ln(C) - \ln(2) - c$
C denotes the (true) mean value of the population (subpopulation) of specific (e.g. fluorescent) molecules of the same kind (the average molecule number) in the observation/detection volume	It describes the analytical sensitivity that the observation/detection volume contains a single molecule of the same kind
Criterion 3	
$p\left(\prod_{i=1}^{\bar{t}} A_i\right)$ <p>Denotes the selfsame molecule likelihood estimators which are the probabilities that a second molecule of the same kind (e.g. a second fluorescent molecule) is outside a boundary at time, meaning outside the lower limit of distance of the observation/detection volume</p>	$1 - \frac{N}{2\Gamma} \cdot \exp\left\{-\frac{\xi^2}{4Dt}\right\} = 1 - \frac{N}{2\Gamma} \cdot \exp\left\{-\frac{\omega_{\sigma}^2}{4DT^*}\right\} \cong 1 - \frac{N}{2\Gamma}$ <p>It shows the likelihood to really see the selfsame molecule that is the individual molecule in the observation/detection volume</p>

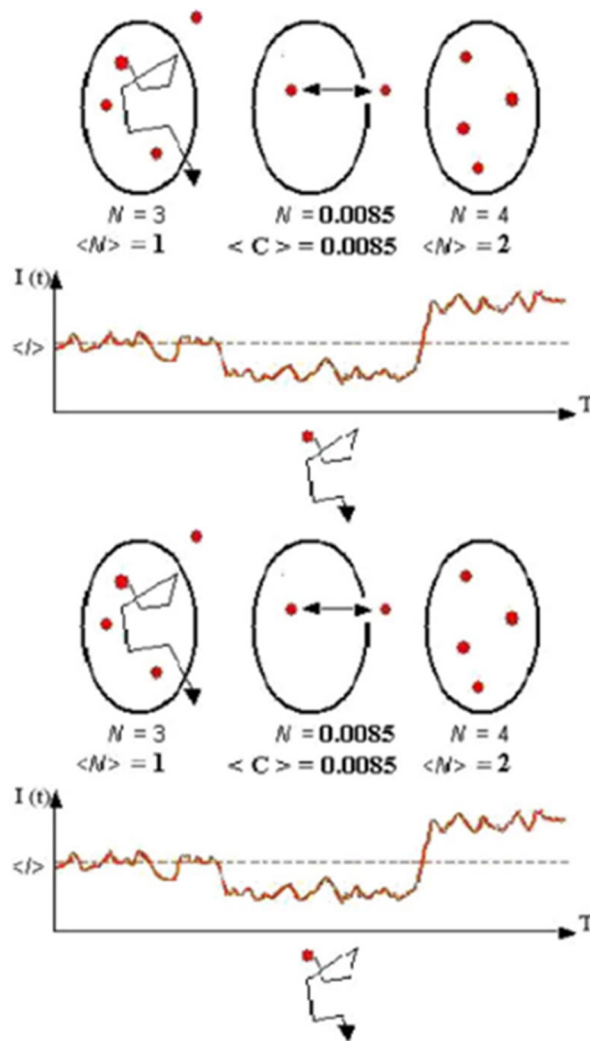


Figure 1: Different levels of abstraction for quantifying thermodynamic jitter

CONCLUSION

Not least because individual molecules (oligonucleotides and single-stranded DNA sequences) were theoretically described in chemical oligonucleotide syntheses on solid supports after release from the solid phase as early as 1994, the biggest breakthrough would be increasing the sensitivity of measurements in liquids (solution) and live cells including membranes without immobilization or hydrodynamic flow. To avoid the single molecule expert in own measurements; use the formulas and relationships provided in this paper. These can be easily and safely applied to determine how many molecules in the measurements are averaged during the measurement of the time.

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